First totally diastereoselective opening of chiral triquinphosphoranes. A new access to enantiopure oxazaphospholidines[†]

Caroline Marchi, Guillaume Delapierre, Frédéric Fotiadu and Gérard Buono*

Laboratoire de Synthèse Asymétrique, UMR 6516, ENSSPICAM, Avenue Escardrille Normandie-Niemen, 13397 Marseille cedex 20, France. E-mail: buono@spi-chim.u-3mrs.fr; Fax: +33 4 91 28 82 47

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Asymmetric addition of isocyanate compounds on chiral triquinphosphoranes, tricyclic hydridophosphoranes, led by a total diastereoselective opening of the diazaphospholidine ring. This provides chiral bicyclic oxazaphospholidines in which an eight-membered ring is fused to the oxazaphospholidine ring by the P–N bond.

Chiral tricoordinated organophosphorus compounds containing an oxazaphospholidine ring have been applied with success in enantioselective catalysis1 and in asymmetric synthesis.2 We wish to report herein an original synthesis of new enantiopure bicyclic oxazaphospholidines based on the reaction between isocyanate compounds and chiral triquinphosphoranes.[‡] This new class of chiral tricyclic phosphoranes was synthesized from enantiopure diaminodiols having a C_2 symmetry axis.³ Those phosphoranes exist as two trigonal bipyramid structures (TBP) with opposite absolute configuration at the phosphorus atom, $R_{\rm P}$ and S_P , in fast equilibrium by a Berry pseudorotation process *via* a SP transition state (Scheme 1).^{3,4} We have described the first asymmetric addition of chiral triquinphosphorane 1c to ketopantolactone and to BH3.SMe2 complex leading to chiral alkoxyphosphorane^{3c} 2 and chiral triquinphosphorane–borane^{3b} adduct 3, respectively. The X-ray diffraction structure of 3 revealed that the borane group coordinated to the axial nitrogen atom of the TBP structure is in the syn position with respect to the P-H bond and in the anti position with respect to the



Scheme 1 Pseudorotation process and addition of isocyanates 4 and 5 on triquinphosphoranes.

† Electronic supplementary information (ESI) available: experimental procedures and characterization of compounds 6a-e, 7a-e, 8. See http: //www.rsc.org/suppdata/cc/b0/b005884j/



adjacent pseudoaxial isopropyl substituent.^{3b,5} Phenyl isocyanate **4** and (*R*)-phenylethyl isocyanate **5** reacted readily with the triquinphosphoranes **1a**– e^6 to afford chiral bicyclic oxazaphospholidines **6a**–e and **7a**–e, respectively, in which an eight-membered ring is fused to the oxazaphospholidine ring by the P–N bond (Scheme 1).

Oxazaphospholidines **6a** and **7a** were formed in quantitative yield by direct reaction of the parent triquinphosphorane **1a** with isocyanates **4** and **5**, respectively, in toluene solution at -30 °C. The exclusive formation of these new compounds could be monitored by ³¹P NMR spectroscopy showing only one downfield singlet at 130.1 ppm for racemic **6a** and two signals at 130.70 and 130.77 ppm for **7a** in a 1:1 ratio corresponding to the diastereomers **7a**- R_P and **7a**- S_P . Due to the high energy barrier for the epimerization of the tricoordinated phosphorus atom (30–35 kcal mol⁻¹),⁷ this result shows that no dynamic chiral discrimination occurred during the attack of chiral isocyanate **5** on the two enantiomeric structures **1a**- R_P and **1a**- S_P in fast equilibrium.

In the case of chiral triquinphosphoranes 1b-e, ³¹P NMR monitoring showed that condensation of isocyanates 4 and 5 occurred with a total diastereoselective opening of the diazaphospholidine ring, to afford a single diastereomer as shown by the ¹³C and ³¹P NMR spectra. Compound 7e reacted with BH3·SMe2 complex in toluene solution to give the borane adduct 8 in which the borane is coordinated to the phosphorus atom (δ ³¹P: 120.3 ppm). The structure of adduct 8 and the absolute configuration of the phosphorus atom were determined by single X-ray diffraction (Fig. 1).8 The nitrogen, oxygen, and boron atoms around the phosphorus center adopt a slightly distorted tetrahedral arrangement with bond angles between 95.0 and 119.8°. Because the complexation of borane occurs with retention of configuration at the phosphorus atom, the borane adduct shows that the diastereoselective opening of the diazaphospholidine ring led to a tricoordinated compound with the S_P absolute configuration. The shortness of the P1–N6 bond [1.637(5) Å instead of 1.67–1.70 Å usually]⁹ can in part be assigned to negative hyperconjugation, due to electron donation from π_N to σ^*_{P-B} orbitals.¹⁰ In fact the N6 nitrogen exhibits a planar configuration [sum of bond angles 359.8(4)°].

The newly formed eight-membered ring adopts a twist-boatchair conformation and the five-membered ring a slightly flattened envelope conformation with the C19 atom as the tip. The benzyl group bound to the C11 carbon of the oxazaphospholidine ring is in the *syn* position with respect to the boron atom whereas the benzyl group bound to the C9 carbon of the eight-membered ring is in the *anti* position.

Taken together, these results suggest that just like borane,^{3b} isocyanates attack preferentially the least hindered axial nitrogen atom of the S_P phosphorane diastereomer (Scheme 2).



Fig. 1 ORTEP drawing of **8**. For more clarity, we omitted hydrogen atoms and used 30% probability ellipsoids. Selected bond lengths [Å]: O2–P1, 1.581(4); O5–P1, 1.592(4); N6–P1, 1.637(5); N4…P, 3.472(4); B23–P1, 1.888(7). Bond angles [°]: B23–P1–O2, 109.3(3); B23–P1–O5, 113.9(3); B23–P1–N6, 119.8(4); N6–P1–O5, 95.0(2); N6–P1–O2, 108.9(2); O2–P1–O5, 109.0(2); P1–N6–C11, 113.6(3); P1–N6–C12, 125.0(3); C8–O2–P1, 122.4(3); C19–O5–P1, 112.3(3); C11–N6–C12, 121.2(3); C9–N4–C13, 119.1(4); C9–N4–C14, 118.2(4); C13–N4–C14, 122.8(5).



Scheme 2 Addition mechanism of isocyanates 4 and 5 on triquinphosphoranes 1b-e.

In fact, the TBP- S_P , the pseudoaxial substituent is in an *anti* position with respect to the lone pair of the axial nitrogen atom whereas it is in a *syn* position in TBP- R_P . The diastereoselectivity observed can thus be rationalized in terms of a kinetically controlled process in which the minor triquinphosphorane diastereomer (29%) reacts faster than the major one (71%) to afford only one diastereomeric oxazaphospholidine.⁴

The importance of steric factors was further confirmed by the fact that the diastereoselectivity of the isocyanate addition depends on the position of the substituents. Indeed, in the case of triquinphosphorane 9,¹¹ in which the phenyl substituents are

 β to the nitrogen atom, the diastereoselectivity of the addition of isocyanates **4** and **5** decrease to 77 and 93%, respectively.

In summary, we disclose for the first time the formation of enantiomerically pure P^{III} oxazaphospholidines from chiral pentacoordinated phosphoranes. These new compounds feature an eight-membered ring, difficult to synthesize with classical methods.¹² Research to evaluate the efficiency of these new compounds in asymmetric catalysis either as ligands or as catalysts is in progress.

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Notes and references

‡ The IUPAC name for triquinphosphorane is hexahydro-11 λ^5 -[1,3,2]ox-azaphospholo[2',3':2,3][1,3,2]diazaphospholo[2,1-*b*][1,3,2]oxazaphosphole.

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- 4 Semi empirical AM1 MO calculations predict that the TBP (R_P) and TBP (S_P) ground-state species are in equilibrium through a SP transition state, the activation barrier being about 5 kcal mol⁻¹. Calculations predict a marked predominance of the (R_P) form [71:29 for **1e**], see ref. 3*b*.
- 5 Compounds **2** and **3** have been studied by X-ray diffraction. They exhibit a percentage of deformation from the ideal TBP structure along the Berry pseudorotation pathway of 66 and 6.9%, respectively.
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- 8 Crystal data for **8**: C₂₉H₃₇BN₃O₈P, $M_r = 517.42$, hexagonal, space group *P*61, *a* = 13.0505(5), *c* = 29.9705(7) Å, *Z* = 6, *V* = 4420.3(3) Å, $\rho_{calcd} = 1.166$ g cm⁻³, $\mu(Mo_{K\alpha}) = 0.126$ cm⁻¹, final *R*1 and *wR*2 are 0.0734 and 0.1745 for 368 parameters and 3328 unique observed reflections with *I* > 2.0 $\sigma(I)$. CCDC 182/1811. See http://www.rsc.org/ suppdata/cc/b0/b005884j/ for crystallographic files in .cif format.
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